## **POSTER PRESENTATION**



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## Mechanistic DFT studies – helicate-type complexes with different alcylic spacers

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*From* 7th German Conference on Chemoinformatics: 25 CIC-Workshop Goslar, Germany. 6-8 November 2011

Metal-controlled self-assembly of complexes is of high interest in the field of Supramolecular Chemistry [1,2]. In the current study, we synthesized binuclear complexes with different spacers and study the influence of chain length on their relative energy. The considered complexes prefer the zigzag conformation. Thus a bridge with an odd number of methylene units forms a meso-Helicate ( $\Delta\Lambda$  or  $\Lambda\Delta$ ) and one with an even number leads to a Helicate ( $\Delta\Delta$  or  $\Lambda\Lambda$ ) (figure 1) [3,4].

Comparison of the calculated transition energies for the non-dissociative interconversions of the diastereomeres with experimental results provides inside into the isomerization process. Moreover, insertion of different cations (templates) into the cavities of the binuclear complexes and corresponding calculations allow prediction of their influence on the isomerization.

Enlargement of the studied system results in binuclear complexes with imino-bridged ligands. The obtained computational results provide a possible explanation for the experimentally observed high diastereoselectivity. As the DFT functionals like B3LYP do not describe long-range interactions properly, we chose the coulombattenuating method CAM-B3LYP [5] which corrects the exchange interaction at long range. The complexes with Ti(IV) in their helical or meso form have been geometrically optimized at the CAM-B3LYP level of theory with the TZVP basis set and MDF10 as ECP for Ti(IV) as implemented in the program package Gaussian09 [6].

Published: 1 May 2012

## References

- 1. Albrecht M: Chem Soc Rev 1998, 27:281.
- 2. Albrecht M, Janser I, Fröhlich R: Chem Commun 2005, 157.
- 3. Albrecht M, Kotila S: Angew Chem 1996, 108:1299.
- 4. Albrecht M, Kotila S: Angew Chem 1995, 107:2285.
- 5. Yanai T, Tew D, Handy N: Chem Phys Lett 2004, 393:51
- 6. Gaussian 09, Revision A02. Gaussian Inc., Wallingford CT; 2009.

doi:10.1186/1758-2946-4-S1-P9

**Cite this article as:** Gossen *et al.*: **Mechanistic DFT studies – helicatetype complexes with different alcylic spacers.** *Journal of Cheminformatics* 2012 **4**(Suppl 1):P9.

Bridge	Form	Energy in a.u.	Nat	1 A
methylene	$\Delta\Delta$	-2519.676584	11XII	1 Alton
methylene	$\Delta \Lambda$	-2519.708944	A SI	1 and
ethylene	$\Delta\Delta$	-2637.656670	4x4	LAFT
ethylene	$\Delta \Lambda$	-2637.637550	417	177
propylene	$\Delta\Delta$	-2755.551314	Y YYYY	2 All
propylene	$\Delta \Lambda$	-2755.570690	1 M	A MIL

Figure 1 Geometrically optimized complexes with Ti(IV); left: energies of the complexes with different alcylic spacers; right:  $\Delta\Delta$  and  $\Lambda\Delta$  form of the complex with a propylene spacer.

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