# DATABASE

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# FluoBase: a fluorinated agents database



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**Abstract** Organofluorine compounds, owing to their unique physicochemical properties, play an increasingly crucial role in fields such as medicine, pesticides, and advanced materials. Fluorinated reagents are indispensable for developing efficient synthetic methods for organofluorine compounds and serve as the cornerstone of organofluorine chemistry. Equally important are fluorinated functional molecules, which contribute specific properties necessary for applications in pharmaceuticals, agrochemicals, and materials science. However, information about these agents' structure, properties, and functions is scattered throughout vast literature, making it inconvenient for synthetic chemists to access and utilize them effectively. Recognizing the need for a dedicated and organized resource, we present FluoBase—a comprehensive fluorinated agents database designed to streamline access to key information about fluorinated agents. FluoBase aims to become the premier resource for information related to fluorine chemistry, serving the scientific community and anyone interested in the applications of fluorine chemistry and machine learning for property predictions. FluoBase is freely available at https://fluobase.sioch emdb.com.

# Scientific contribution

FluoBase is a database designed to provide comprehensive information on the structures, properties, and functions of fluorinated agents and functional molecules. FluoBase aims to become the premier resource for fluorine chemistry, serving the scientific community and anyone interested in the applications of fluorine chemistry and machine learning for property predictions.

Keywords Organofluorine chemistry, Database, Reagent, Agent

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# Introduction

Due to their unique biological, physical, and chemical properties, organofluorine compounds are becoming increasingly important in various fields of chemistry and everyday life [1-3]. For instance, fluorine has emerged as a "magic element" in pharmaceuticals and agrochemicals, with approximately 30% of all agrochemicals and 20% of pharmaceuticals on the market containing fluorine [4–11]. This is due to the fact that the incorporation of fluorine or fluoroalkyl groups into drug molecules often enhances pharmacokinetic properties, binding selectivity, and metabolic stability, thereby contributing to their overall efficacy and stability. Additionally, fluorinated compounds are extensively utilized in many advanced functional materials [12–17]. This growing utilization has led to an increased demand for safe and efficient reagents and selective synthetic methodologies for introducing fluorine or fluorine-containing groups into organic compounds [18-33].

Over the past few decades, diverse fluorinating and fluoroalkylating reagents have been developed [34–50], significantly advancing synthetic fluorine chemistry. However, the details regarding the structure, properties, and functions of these reagents, as well as fluorinated functional molecules, are scattered across numerous sources, presenting a challenge for synthetic chemists seeking to efficiently integrate this information into their work.

In light of the growing significance of databases in the pharmaceutical and chemical sciences-particularly with the advent of artificial intelligence [51–56], establishing a dedicated open-source database for fluorinated agents presents a valuable opportunity. Existing resources, such as PubChem [57, 58], ChEMBL database [59], ZINC database [60], Protein Data Bank [61], DrugBank [62],

Open Reaction Database [63], *i*Bond [64], database for natural products [65–67] and so on, underscore the benefits of centralized information.

In this contribution, we wish to report a comprehensive and publicly accessible database, FluoBase, dedicated to fluorinated agents. FluoBase will provide detailed information on the physical and chemical characteristics and synthetic methods of various fluorinating reagents. By offering this valuable and easily accessible resource, FluoBase aims to foster the development and application of fluorinated reagents. In addition to fluorinating reagents, FluoBase will also encompass a wide range of fluorinated functional molecules, including natural products, refrigerants, anesthetics, surfactants, and drug molecules. Launched in 2024, FluoBase aspires to become the premier resource for information related to fluorine chemistry, serving the scientific community and anyone interested in the applications of fluorine chemistry and machine learning for property predictions.

## **Construction and content**

The main objective of this work is to accelerate research and stimulate new discoveries in fluorine chemistry. To achieve this, a robust and efficient database was developed using MongoDB, a document database designed for modern application development and cloud environments, storing data in BSON (Binary JSON) format, which allows for a flexible schema. This means documents with different structures can be stored within the same collection. Despite MongoDB's non-relational nature and lack of foreign keys, an ID system was implemented in the *reference* collection. This ID was then used within the *reagent* collection as a pseudo foreign key, enabling each dataset to be linked back to its source in the literature. This design facilitates easy verification of data origins by allowing users to access the corresponding journal page directly through provided links.

In addition to the curated collections, the database integrates a wide range of comprehensive data sourced from scientific literature. This includes detailed information on 2050 molecules (comprising 1810 fluorinated reagents and 240 fluorinated functional molecules), 7373 properties, 747 3D structures, 200 CCDC crystal data and 1208 references spanning all these datasets. This wealth of information provides researchers with an invaluable resource to explore, analyze, and advance the field of fluorine chemistry.

To facilitate efficient interaction with the database, both GraphQL and REST APIs were implemented. The REST API is primarily used for retrieving images from SMILES representations, while GraphQL handles all other data operations, enabling more flexible and efficient queries. These powerful developer tools not only simplify the process of evolving APIs over time but also offer a complete and unified view of the data within the database (Fig. 1). Furthermore, the APIs allow for seamless integration with artificial intelligence systems, making the database a valuable resource for predicting the properties of fluorinated compounds.

Furthermore, the database offers carefully curated information on a wide range of fluorinated compounds. Key data, such as molecular structures and properties, are easily accessible through well-organized features, with regular updates ensuring that the database remains a reliable and up-to-date resource for the scientific community.

#### Utility and discussion

We have developed a dedicated web page for the Fluorinated Agent Database, enabling users to search for information directly from its homepage (refer to Fig. 2). Users can find desired data by entering criteria such as name in Chinese or English, chemical formula, CAS Registry Number (CASRN), abbreviation (Abbr), DOI, or SMILES into the search bar. This user-friendly interface ensures efficient access to comprehensive information on fluorinated reagents (with support for additional agents in the future), enhancing usability for researchers and practitioners alike. This database is also available in two languages. By default, the language is set according to the browser settings: Chinese remains in Chinese, while all other languages default to English. Additionally, in the navigation bar you can choose between English and Chinese. Given that these are the most widely spoken languages globally, this option enhances the database's accessibility for a larger audience.

## **Fluorinated reagents**

The database contains extensive information on a wide variety of fluorinated reagents, encompassing more than twenty distinct groups. Figure 3 highlights the largest group, each containing over 50 reagents. In addition, there are many other fluorinated reagents in the database, but as there are not so many, it was decided not to include them in the Fig. 3, but to write them down. Below all possible reagents, which are not in the Fig. 3, have been listed: *fluorinated reagents*,



Fig. 1 Workflow of FluoBase



Fig. 2 Main page of FluoBase



# **Data Distribution for Fluorinated Reagents**

Fig. 3 Data distribution for fluorinated reagents in FluoBase

monofluoromethylating, monofluoromethylthiolating, difluoromethylthiolating, trifluoromethylsulfinylatrifluoromethylsulfonylation, fluorosulfonylation, tion,

fluoroalkylthiolation, monofluoromethoxylating, monofluoromethoxylating, fluoroalkoxylation, trifluoromethylaminating, fluoroalkylamination, fluoroalkylselenylation,





Fig. 4 a The proportion of types for fluorinating reagents. b The proportion of types for trifluoromethylating reagents

fluoroalkyltellurylation, pentafluorosul, fanylation, difluorocarbene and fluoroalkyl carbenes reagents.

Furthermore, the fluorinated reagents have been categorized into three main classes: nucleophilic, electrophilic and radical. Examples of these categories are illustrated in Fig. 4a, b. It is noteworthy that in cases where the electrophilic class is not absent, the amount of this electrophilic class is the most numerous one or almost equal to another group. Such situation occurs for almost all other fluorinated reagents except for *trifluoromethoxylating* reagents and fluoroalkylation reagents.

These data can be got by using search bar, as previously mentioned, or by utilizing the Ketcher tool [68] (by clicking picture of pen next to magnifying glass icon). The Ketcher tool within the interface allows user to draw molecules and then convert them into SMILES format. This SMILES format is then used to search for data within the Fluorinated Agent Database. In addition, the Ketcher tool made it possible to obtain a molecule image of the base of the SMILES representation. Apart from Ketcher, they are few more tools which can be used. For example, it is possible to mark favourite reagents to be remembered. Navigation is facilitated through the top navigation bar, where users can select desired reagents. After pressing the button inside the navigation bar, the desired reagents will be displayed in the next open page. Then on the left side it is possible to use filter. Filter contains possible options to choose like "Crystal Information", "3D Structure", "Homolytic BDE", "Heterolytic BDE" or "Redox Potential" (Fig. 5). After selected options, "Submit" button can be pressed and desired reagents will appear. In addition to the internal options of the filter, it is possible to select the range, like for example "Enable HBDE range" or "Enable Redox range" (Fig. 5b). Then on the top can be selected "All", "Nucleophilic", "Electrophilic" or "Radical" (Fig. 5c). In addition, by clicking on the pencil icon next to the magnifying glass icon in the search bar, it is possible to draw a chemical structure and use it to search for the desired results. If result was not found then there is small tool called "structure similar search". This tool also evaluates similarity and show it under the figure (for instance: "Similarity: 0.74").

Result page contains basic information like for example: name, formula, InChI, InChI Key, canonical SMILES and link to reference (Please visit our website for more details). It is then possible to check properties such as: appearance, density, boiling point, melting point, refractive index, solubility, stability, Mayr's electrophilicity/nucleophilicity parameter [69, 70] and homolytic or heterolytic BDE (Please visit our website for more details). Besides result page contains synthetic method and 3D structure (Fig. 6), which is particularly helpful for young scientists seeking guidance. 3D structures are sourced from crystallographic data when available; if not, PubChem data is used [71], and in the absence of both, no 3D structure is provided. Users can also copy data and save images in svg format.

### Other fluorinated agents

In addition to comprehensive information on a wide range of fluorinated reagents, FluoBase also aims to provide detailed data on a broad array of fluorinated functional molecules such as natural products, refrigerants, anesthetics and surfactants (Fig. 7). This database is constantly being improved and some new agents like drug



Fig. 5 Three classes possible to choose from top menu: a View from the filter. b Filter for Homolytic BDE range and Heterolytic BDE. c Filter for reagent type



Fig. 6 a View of the synthetic method. b View of the 3D structure

molecules, insulating gas are under construction and will

The search for substances can also be facilitated by

the top navigation bar, where the user can select the

substances of interest. The second option to choose

be available soon.

desired fluorinated agent is to press image on the main page in section "Agent" which will send user directly to the page containing agents. The possible agents that can be reviewed are: natural product, refrigerant, anesthetic and surfactant. All of the above can be selected from the



Fig. 7 Data distribution for fluorinated agents in FluoBase

navigation bar. Below is an example of the results page for Agent—Natural Product. In Fig. 8a there are fluorinated natural products. Clicking on the first image (24751-69-7) opens the details page (Fig. 8b). The subpage is divided into three sections. The first section called Basic Information contains details such as formula, abbreviation, Chinese name, exact mol weight, InChI, other name or CAS register number. The second section called property contains details such as density, boiling point, bioactivity. The last section called biological source contains data such as reference. Clicking on this reference will take you to an external page with the paper used to obtain the data. On the left is a vertical navigation bar that runs along the side of a page and shows all the links that will take the user to different parts of a page.

### Submit your own data

Additionally, users have the capability to submit their own data to the Fluorinated Agent Database, including references, reagents, and files in "SDF" and "CIF" formats for 3D structures. The whole procedure for adding new data has been divided into five steps. All steps have to be prepared to finish the procedure, but user in every moment can return to the previous step. In the first four steps the user can add his own data such as: smiles, name, CSD Doi, CCDC No (Fig. 9a); density, electrophilicity, stability, solubility (Fig. 9b); reaction references, reaction method, reaction smiles (Fig. 9c); type reference, redox (redox potential) reference, bde (bond dissociation energy) reference, hbde (heterolytic bond dissociation energy) reference (Fig. 9d). Furthermore, there is an option to check if the input for the drawing is correct, if it exists in the database and then there is a possibility to obtain canonical smiles. Once all four steps have been well prepared, the fifth and final step allows the user to update the reference and send their data. Once the data has been sent, there is no way of going back, so the user should prepare his data very carefully. The data provided must first undergo a verification process. In the event of a problem, the user will be informed of any errors, ambiguities or missing data. If there is no problems then obviously new data will be approved.

However, to access this feature, users must first log in to the system. If user does not have account, it is possible to create one by providing data like username, institution, email and password. After provide captcha from received email it is possible to sign in and submit own data. Later user can update his details in the: User Info page. Registering for an account is free of charge and takes only a short amount of time to complete (Fig. 10). Then using FluoBase is free of charge and is intended for non-profit academic purposes only. Users of the database are kindly requested to register and cite the website: https://fluobase.siochemdb.com, as well as the original literature in their work.





Fig. 8 a View of the fluorinated natural products. b View of the details for 4'-Fluoro-5'-Osulfamoyaladenosine



Fig. 9 Pages for data submission. a Step 1. b Step 2. c Step 3. d Step 4

	Institution
2 JaneDoe	Your institution
5 to 15 characters. Letters, digits and @, ., +, -, _ or	y. Your institution or organization, optional
Password	Confirm password
Here Your password	Confirm password
8 to 16 characters. At least one uppercase letter, lowercase letter, and number.	Confirm your password
Email	Captcha
jane@example.com	•
The email used for password reset	Email Captcha, case-insensitive
<ul> <li>→ Slide Right to Fill Puzzle</li> <li>Register</li> </ul>	Send Email

Fig. 10 Page for registration

# Machine Learning Predictions of <sup>19</sup>F NMR chemical shifts

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Brief Introduction to the <sup>19</sup>F NMR chemical shifts Prediction Tool:

1. Purpose and scope: This tool predicts<sup>19</sup>F NMR chemical shifts for most kinds of organic molecules.

2. Prediction methods: Graph Convolutional Network (GCN) with root-mean-square error (RMSE) = 5.019 ppm, R<sup>2</sup> = 0.991 and mean absolute errors (MAE) = 3.636 ppm.

3. Dataset: The dataset for training contains about 3376 unique experimental <sup>19</sup>F NMR chemical shifts

#### Please Cite:

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Fig. 11 Machine learning prediction of <sup>19</sup>F NMR chemical shifts

# <sup>19</sup>F NMR prediction tool

The last option in the navigation bar is not a link to the next set of data, but rather a tool called the "<sup>19</sup>F NMR Prediction Tool." This tool enables users to predict the <sup>19</sup>F NMR chemical shifts for most types of organic molecules [72]. The training dataset consists of approximately 3,376 unique experimental <sup>19</sup>F NMR chemical shifts. After entering a SMILES string in the search bar or drawing a structure using Ketcher, the <sup>19</sup>F NMR Prediction Tool employs a Graph Convolutional Network (GCN) as its prediction method to generate the results. This model demonstrates high accuracy, with a root mean square error (RMSE) of 5.019, a coefficient of determination (R<sup>2</sup>) of 0.991, and a mean absolute error (MAE) of 3.636 ppm (Fig. 11).

#### **Future plans**

In the near future, the database will be expanded to include fluorinated functional molecules, such as insulating gases, surfactants, drug molecules, and more. Furthermore, plans are underway to enrich the database with additional data, including key pharmacologically relevant properties such as logP/logD and water solubility. New tools will also be integrated to enable users to quickly locate desired information and analyze the data more efficiently.

#### Conclusions

In this study, we introduce FluoBase, a publicly accessible database dedicated to providing comprehensive information on the physical and chemical properties of various fluorinating agents. FluoBase will cover a wide range of fluorinated substances, including refrigerants, anesthetics, surfactants, and drug molecules. Committed to ongoing development, FluoBase will regularly introduce new services, tools, and enhancements to existing features. Systematic updates will ensure the incorporation of the latest advancements in fluorine chemistry. As a key platform for fluorine chemistry, FluoBase aims to serve the scientific community and anyone interested in the applications of fluorinating agents and machine learning for property prediction.

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# Abbreviations

Abbreviations		
	REST	Representational state transfer
	API	Application Programming Interface
	CAS	Chemical Abstracts Service
	InChl	International Chemical Identifier
	SMILES	Simplified Molecular Input Line Entry System
	CCDC	The Cambridge Crystallographic Data Centre
	DOI	Digital Object Identifier
	CIE	Crystallographic Information File

\_IF Crystallographic Information File

- SDF Structured Data File
- BDE Bond Dissociation Energy

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Not applicable.

#### Author contributions

R.M., Y.L. and X.S.X. coordinated the writing of the paper. Y.L. designed and developed the FluoBase database. D.S., W.S.H., L.Z., H.H., X.L. contributed to data collection. Y.L. and X.S.X. conceived and supervised the project. All authors actively contributed to the FluoBase database and have read and approved the final manuscript.

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#### Availiability of data materials

No datasets were generated or analysed during the current study.

#### Declarations

#### **Competing interests**

The authors declare no competing interests.

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#### References

- 1. Gouverneur V, Seppelt K (2015) Introduction: fluorine chemistry. Chem Rev 115(2):563–565
- Prakash GKS, Wang F, O'Hagan D, Hu J, Ding K, Dai L-X (2012) Flourishing frontiers in organofluorine chemistry. In: Ding K, Dai L-X (eds) Organic chemistry—breakthroughs and perspectives 12. Wiley-VCH Verlag & Co KGaA, Weinheim, pp 413–476
- Ni C, Hu J (2016) The unique fluorine effects in organic reactions: recent facts and insights into fluoroalkylations. Chem Soc Rev 45(20):5441–5454
- Gillis EP, Eastman KJ, Hill MD, Donnelly DJ, Meanwell NA (2015) Applications of fluorine in medicinal chemistry. J Med Chem 58(21):8315–8359
- Jeschke P (2004) The unique role of fluorine in the design of active ingredients for modern crop protection. ChemBioChem 5(5):570–589
- Zhou Y, Wang J, Gu Z, Wang S, Zhu W, Aceña JL, Soloshonok VA, Izawa K, Liu H (2015) Next generation of fluorine-containing pharmaceuticals, compounds currently in phase II-III clinical trials of major pharmaceutical companies: new structural trends and therapeutic areas. Chem Rev 116(2):422–518
- Ogawa Y, Tokunaga E, Kobayashi O, Hirai K, Shibata N (2018) Current contributions of organofluorine compounds to the agrochemical industry. IScience 23(9):1–23
- Meanwell NA (2018) Fluorine and fluorinated motifs in the design and application of bioisosteres for drug design. J Med Chem 61(14):5822–5880
- Shabir G, Saeed A, Zahid W, Naseer F, Riaz Z, Khalil N, Muneeba AF (2023) Chemistry and pharmacology of fluorinated drugs approved by the FDA (2016–2022). Pharmaceuticals 16(8):1162
- Inoue M, Sumii Y, Shibata N (2020) Contribution of organofluorine compounds to pharmaceuticals. ACS Omega 5(19):10633–10640
- Johnso BM, Shu YZ, Zhuo X, Meanwell NA (2020) Metabolic and pharmaceutical aspects of fluorinated compounds. J Med Chem 63(12):6315–6386

- 12. Berger R, Resnati G, Metrangolo P, Weber E, Hulliger J (2011) Organic fluorine compounds: a great opportunity for enhanced materials properties. Chem Soc Rev 40(7):3496–3508
- Zhang C, Yan K, Fu C, Peng H, Hawker CJ, Whittaker AK (2022) Biological utility of fluorinated compounds: from materials design to molecular imaging. Ther Environ Remed Chem Rev 122(1):167–208
- Dolui S, Kumar D, Banerjee S, Ameduri B (2021) Well-Defined Fluorinated Copolymers: Current Status and Future Perspectives. Accounts Mater Res 2(4):242–251
- Lemoine K, Hémon-Ribaud A, Leblanc M, Lhoste J, Tarascon JM, Maisonneuve V (2022) Fluorinated materials as positive electrodes for Li- and Na-ion batteries. Chem Rev 122(18):14405–14439
- Zhang Z, Chen K, Ameduri B, Chen M (2023) Fluoropolymer nanoparticles synthesized via reversible-deactivation radical polymerizations and their applications. Chem Rev 123(22):12431–12470
- Wang Y, Yang X, Meng Y, Wen Z, Han R, Hu X, Sun B, Kang F, Li B, Zhou D, Wang C, Wang G (2024) Fluorine chemistry in rechargeable batteries: challenges, progress, and perspectives. Chem Rev 124(6):3494–3589
- Chu L, Qing FL (2014) Oxidative trifluoromethylation and trifluoromethylthiolation reactions using (trifluoromethyl)trimethylsilane as a nucleophilic CF<sub>3</sub> source. Acc Chem Res 47(5):1513–1522
- Sap JBI, Meyer CF, Straathof NJW, Iwumene N, Am Ende CW, Trabanco AA, Gouverneur V (2021) Late-stage difluoromethylation: concepts, developments and perspective. Chem Soc Rev 50(14):8214–8247
- Butcher TW, Amberg WM, Hartwig JF (2022) Transition-metal-catalyzed monofluoroalkylation: strategies for the synthesis of alkyl fluorides by C–C bond formation. Angew Chem Int Ed 61(7):e202112251
- Sheldon DJ, Crimmin MR (2022) Repurposing of F-gases: challenges and opportunities in fluorine chemistry. Chem Soc Rev 51(12):4977–4995
- 22. Lou TSB, Willis MC (2022) Sulfonyl fluorides as targets and substrates in the development of new synthetic methods. Nat Rev Chem 6:146–162
- Pons A, Delion L, Poisson T, Charette AB, Jubault P (2021) Asymmetric synthesis of fluoro, fluoromethyl, difluoromethyl, and trifluoromethylcyclopropanes. Acc Chem Res 54(14):2969–2990
- 24. Ma X, Song Q (2020) Recent progress on selective deconstructive modes of halodifluoromethyl and trifluoromethyl-containing reagents. Chem Soc Rev 49(24):9197–9219
- Mykhailiuk PK (2021) Fluorinated pyrazoles: from synthesis to applications. Chem Rev 121(3):1670–1715
- 26. Xiao P, Pannecoucke X, Bouillon JP, Couve-Bonnaire S (2021) Wonderful fusion of organofluorine chemistry and decarboxylation strategy. Chem Soc Rev 50(10):6094–6151
- 27. Xiao H, Zhang Z, Fang Y, Zhu L, Li C (2021) Radical trifluoromethylation. Chem Soc Rev 50(11):6308–6319
- Cabré A, Verdaguer X, Riera A (2022) Recent advances in the enantioselective synthesis of chiral amines via transition metal-catalyzed asymmetric hydrogenation. Chem Rev 122(1):269–339
- Pratley C, Fenner S, Murphy JA (2022) Nitrogen-centered radicals in functionalization of Sp2Systems: generation, reactivity, and applications in synthesis. Chem Rev 122(9):8181–8260
- Ma J-A, Cahard D (2008) Update 1 of: asymmetric fluorination, trifluoromethylation, and perfluoroalkylation reactions. Chem Rev 108(9):PR1–PR43
- 31. Campbell MG, Ritter T (2015) Modern carbon-fluorine bond forming reactions for aryl fluoride synthesis. Chem Rev 115(2):612–633
- Champagne PA, Desroches J, Hamel JD, Vandamme M, Paquin JF (2015) Monofluorination of organic compounds: 10 years of innovation. Chem Rev 115(17):9073–9174
- 33. Zhu Y, Han J, Wang J, Shibata N, Sodeoka M, Soloshonok VA, Coelho JAS, Toste FD (2018) Modern approaches for asymmetric construction of carbon-fluorine quaternary stereogenic centers: synthetic challenges and pharmaceutical needs. Chem Rev 118(7):3887–3964
- Langlois BR, Billard T, Roussel S (2005) Nucleophilic trifluoromethylation some recent reagents and their stereoselective aspects. J Fluorine Chem 126(2):173–179
- Si Y, Tang P (2023) Development and application of trifluoromethoxylating reagents. Chinese J Chem 41(17):2179–2196
- Wu J, Shen Q (2021) Difluoromethylthiolator: a toolbox of reagents for difluoromethylthiolation. Acc Chem Res 54(14):2946–2958

- Li M, Xue X-S, Cheng J-P (2020) Establishing cation and radical donor ability scales of electrophilic F, CF<sub>3</sub>, and SCF<sub>3</sub> transfer reagents. Acc Chem Res 53(1):182–197
- Shibata N, Matsnev A, Cahard D (2010) Shelf-stable electrophilic trifluoromethylating reagents: A brief historical perspective. Beilstein J Org Chem 6:65
- Umemoto T (1996) Electrophilic perfluoroalkylating agents. Chem Rev 96(5):1757–1778
- Charpentier J, Früh N, Togni A (2015) Electrophilic trifluoromethylation by use of hypervalent iodine reagents. Chem Rev 115(2):765–825
- Ni C, Hu M, Hu J (2014) Good partnership between sulfur and fluorine: sulfur-based fluorination and fluoroalkylation reagents for organic synthesis. Chem Rev 115:765–825
- Qing F-L, Liu X-Y, Ma J-A, Shen Q, Song Q, Tang P (2022) A fruitful decade of organofluorine chemistry: new reagents and reactions. CCS Chem 4(8):2518–2549
- Umemoto T, Yang Y, Hammond GB (2021) Development of N-F fluorinating agents and their fluorinations: historical perspective. Beilstein J Org Chem 27(17):1752–1813
- Meyer D, Jangra H, Walther F, Zipse H, Renaud P (2018) A third generation of radical fluorinating agents based on N-fluoro-N-arylsulfonamides. Nat Commun 9:4888
- Neumann CN, Ritter T (2017) Facile C–F bond formation through a concerted nucleophilic aromatic substitution mediated by the PhenoFluor reagent. Acc Chem Res 50(11):282–2833
- Shibata N (2016) Development of shelf-stable reagents for fluoro-functionalization reactions. Bull Chem Soc Jpn 89(11):1307–1320
- Umemoto T (2014) Exploration of fluorination reagents starting from FITS reagents. J Fluorine Chem 167:3–15
- Lin JH, Xiao JC (2020) Fluorinated ylides/carbenes and related intermediates from phosphonium/sulfonium salts. Acc Chem Res 53(8):1498–1510
- 49. Xie Q, Hu J (2024) A journey of the development of privileged difluorocarbene reagents  $TMSCF_2X$  (X = Br, F, Cl) for organic synthesis. Acc Chem Res 57(5):693–713
- Reichel M, Karaghiosoff K (2020) Reagents for selective fluoromethylation: a challenge in organofluorine chemistry. Angew Chem Int Ed 59(30):12268–12281
- Yang X, Wang Y, Byrne R, Schneider G, Yang S (2019) Concepts of artificial intelligence for computer-assisted drug discovery. Chem Rev 119(18):10520–10594
- Paul D, Sanap G, Shenoy S, Kalyane D, Kalia K, Tekade RK (2021) Artificial intelligence in drug discovery and development. Drug Discov Today 26(1):80–93
- Liu Y, Yang Q, Li Y, Zhang L, Luo S (2020) Application of machine learning in organic chemistry. Chin J Org Chem 40(11):3812–3827
- Dou B, Zhu Z, Merkurjev E, Ke L, Chen L, Jiang J, Zhu Y, Liu J, Zhang B, Wei GW (2023) Machine learning methods for small data challenges in molecular science. Chem Rev 123(13):8736–8780
- Meuwly M (2021) Machine learning for chemical reactions. Chem Rev 121(16):10218–10239
- Hong X, Yang Q, Liao K et al (2024) Al for organic and polymer synthesis. Sci China Chem 67:2461–2496
- Kim S, Thiessen PA, Bolton EE, Chen J, Fu G, Gindulyte A, Han L, He J, He S, Shoemaker BA, Wang J, Yu B, Zhang J, Bryant SH (2016) PubChem substance and compound databases. Nucleic Acids Res 44(D1):D1202–D1213
- Kim S, Chen J, Cheng T, Gindulyte A, He J, He S, Li Q, Shoemaker BA, Thiessen PA, Yu B, Zaslavsky L, Zhang J, Bolton EE (2021) PubChem in 2021: new data content and improved web interfaces. Nucleic Acids Res 49(D1):D1388–D1395
- Bento AP, Gaulton A, Hersey A, Bellis LJ, Chambers J, Davies M, Kruüger FA, Light Y, Mak L, McGlinchey S, Nowotka M, Papadatos G, Santos R, Overington JP (2014) The ChEMBL bioactivity database: an update. Nucleic Acids Res 42(D1):D1083–D1090
- Irwin JJ, Shoichet BK (2005) ZINC–a free database of commercially available compounds for virtual screening. J Chem Inf Model 45(1):177–182
- Berman HM, Westbrook J, Feng Z, Gilliland G, Bhat TN, Weissig H, Shindyalov IN, Bourne PE (2000) The protein data bank. Nucleic Acids Res 28(1):235–242

- 62. Wishart DS, Knox C, Guo AC, Shrivastava S, Hassanali M, Stothard P, Chang Z, Woolsey J (2006) DrugBank: a comprehensive resource for in silico drug discovery and exploration. Nucleic Acids Res 34(suppl1\_1):D668–D672
- Kearnes SM, Maser MR, Wleklinski M, Kast A, Doyle AG, Dreher SD, Hawkins JM, Jensen KF, Coley CW (2021) The open reaction database. J Am Chem Soc 143(45):18820–18826
- 64. Yang J-D (2020) Internet bond-energy databank (pK<sub>a</sub> and BDE)–iBonD Home Page. http://ibond.nankai.edu.cn.
- Banerjee P, Erehman J, Jiansong F, Gohlke B-O, Wilhelm T, Dunkel PRM (2015) Super Natural II–a database of natural products. Nucleic Acids Res 43(D1):D935–D939
- Xu T, Chen W, Zhou J, Dai J, Li Y, Zhao Y (2020) NPBS database: a chemical data resource with relational data between natural products and biological sources. Database 2020:baaa102
- Sorokina M, Merseburger P, Rajan K, Yirik MA, Steinbeck C (2021) COCO-NUT online: collection of open natural products database. J Cheminform 13(1):2
- Karulin B, Kozhevnikov M (2011) Ketcher: web-based chemical structure editor. J Cheminform 3(Suppl 1):P3
- Mayr H, Patz M (1994) Scales of nucleophilicity and electrophilicity: a system for ordering polar organic and organometallic reactions. Angew Chem Int Ed 33(9):938–957
- Mayr H, KempfArmin B, Ofial R (2003) π-Nucleophilicity in carbon-carbon bond-forming reactions. Acc Chem Res 36(1):66–77
- Bolton EE, Chen J, Kim S, Han L, He S, Shi W, Simonyan V, Sun Y, Thiessen PA, Wang J, Yu B, Zhang J, Bryant SH (2011) PubChem3D: a new resource for scientists. J Cheminform 3(1):32
- Li Y, Huang W-S, Zhang L, Su D, Xu H, Xue X-S (2024) Prediction of <sup>19</sup>F NMR chemical shift by machine learning. Artif Intell Chem 2(1):100043

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